

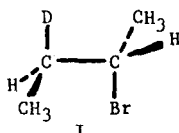
STEREOCHEMISTRY OF BASE-CATALYZED ELIMINATIONS FROM 2-ALKYL BROMIDES

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A syn-elimination stereochemistry has been postulated in eliminations from 2-hexyl chloride, bromide, and iodide induced by t-BuOK/DMSO on the basis of the very high trans-/cis-2-hexene ratios observed (1). A stereochemical study of eliminations from 2-alkyl halides exhibiting high trans-/cis-2-alkene ratios (i.e., promoted by t-BuOK/DMSO and n-Bu₄NF/DMF (2)) seemed warranted. We report the results of such an investigation using erythro-3-deutero-2-bromobutane, I.



I was prepared by the procedure of Skell and Hall (3) with the modification of keeping the reaction mixture temperature between -95° and -85° during irradiation. The presence of approximately 3% of contaminating threo-isomer was indicated by 6% of trans-2-butene-d in the trans-2-butene formed in reactions of I with KOEt/EtOH (Table 2).

I and undeuterated 2-butyl bromide, II, were subjected to elimination reactions in three different base/solvent systems. The relative olefinic proportions are presented in Table 1. The k_H/k_D values in Table 1 are calculated from the product percentages (4).

The k_H/k_D values show that deuterium was not removed during formation of cis-2-butene, but deuterium loss was involved in formation of trans-2-butene. The isotopic analysis of the olefins (Table 2) supports this conclusion. The data demonstrate quite clearly that both cis- and trans-2-butene are formed by anti-elimination in reactions of I with t-BuOK/DMSO and n-Bu₄NF/DMF.

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TABLE 1

Olefin Proportions and k_H/k_D Values from Elimination of erythro-3-Deutero-2-bromobutane

Reaction Conditions	Reactant	1-Butene %	<u>trans</u> -2-Butene ^a %	k_H/k_D	<u>cis</u> -2-Butene %	k_H/k_D
EtOK/EtOH (70°)	I	35.8 ^b	29.4	3.4	34.8	1.0
	II	20.8	58.6	---	20.6	---
<u>t</u> -BuOK/DMSO (30°)	I	53.3	21.0	4.5	25.7	1.1
	II	29.4	55.0	---	15.5	---
<u>n</u> -Bu ₄ NF/DMF (50°)	I	33.1	33.1	3.6	33.8	1.1
	II	16.7	64.5	---	18.7	---

^a Corrected for presence of trans-2-butene-d.^b These olefinic proportions agree with those given in reference 3.

TABLE 2

Isotopic Composition^{a, b} of Olefins from Elimination of erythro-3-Deutero-2-bromobutane

	EtOK/EtOH %	<u>t</u> -BuOK/DMSO %	<u>n</u> -Bu ₄ NF/DMF %
1-butene-d	100	100	100
1-butene	0	0	0
<u>trans</u> -2-butene-d	6	6	6
<u>trans</u> -2-butene	94	94	94
<u>cis</u> -2-butene-d	100	100	100
<u>cis</u> -2-butene	0	0	0

^a Measured by mass spectrometry at low ionizing voltage.^b Uncertainty estimate as $\pm 1\%$.

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